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Charge-compensation models of four Cr^{3+} and four Fe^{3+} centres in KTiOPO₄

Sang Won Ahn and Sung Ho Choh[†]

Department of Physics, Korea University, Seoul 136-701, Republic of Korea

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Abstract. An overall comparison between the principal-axis orientations of the **g**-tensors and the second-order zero-field-splitting (ZFS) tensors for four Cr^{3+} and four Fe^{3+} centres reported on in our previous studies has been made. One direction of the principal axes of the **g**-tensor is nearly the same as one of those of the second-order ZFS tensor for Cr^{3+} centres, but a similar direction does not exist for Fe^{3+} centres. It is also found that one direction of the principal axes of the second-order ZFS tensors for Cr^{3+} centres. It is suggested that Cr^{6+} ions substituted for P^{5+} in PO₄ tetrahedra play the role of positive monovalent charge compensators for both Cr^{3+} and Fe^{3+} , replaced at the sites Ti(1) and Ti(2). Possible charge-compensation models related to Cr^{6+} ions for providing an explanation of the origins of the two different centres arising from the Cr^{3+} and Fe^{3+} at crystallographically equivalent Ti sites are also suggested.

1. Introduction

Potassium titanyl phosphate KTiOPO₄ (KTP) is a new kind of non-linear optical crystal having a high optical damage threshold [1, 2]. As a result this crystal has been widely applied in areas ranging from frequency doublers [1–4], modulators and Q-switches [5, 6], and low-loss waveguide fabrication [7, 8] to piezoelectric applications [9]. For the last ten years, there have been extensive EPR studies of the crystals carried out in order to provide meaningful information on the microscopic structures around point defects in KTP. KTP crystallizes in the orthorhombic space group $Pna2_1$ with eight formulae per unit cell [10, 11]. The structure exhibits two crystallographically different Ti sites: Ti(1) and Ti(2), replaced by Cr³⁺ and Fe³⁺. Each Ti has four chemically equivalent but magnetically inequivalent sites [12–15] due to the symmetry elements, characterized by two glide planes *n* and *a* and one screw axis 2_1 [10, 11].

On the basis of previous work [16–19], we were able to identify four Cr^{3+} [12, 13] and four Fe³⁺ [14, 15] centres, and all four complete sets of fine structures due to the ions at the magnetically inequivalent sites were systematized in terms of the crystal symmetry of KTP. In this study, an overall comparison between the principal-axis orientations of the **g**-tensors and the second-order zero-field splitting (ZFS) tensors for Cr^{3+} and Fe³⁺ centres is made. Possible charge-compensation models related to Cr^{6+} substituted for P⁵⁺ in PO₄ tetrahedra for providing an explanation of the origins of the two different centres arising from Cr^{3+} and Fe³⁺ at two crystallographically equivalent Ti sites are suggested.

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[†] Author to whom any correspondence should be addressed. E-mail address: shchoh@kuccnx.korea.ac.kr; fax: 82-2-927-3292; telephone: 82-2-3290-3091.

2. Experiments

For the EPR measurements two kinds of KTP crystal were synthesized by the flux method using the chemical $K_6P_4O_{13}$ [20], known as a particularly suitable flux for growth of KTP crystal [21]. One is nominally undoped and the other is doped with 0.01 mol% of Fe. The Cr^{3+} ions in the undoped crystal were present as an impurity in the starting material and were identified by comparing our EPR results [12, 13] with those given in reference [18].

The EPR measurements were carried out at room temperature by employing a Bruker Q-band spectrometer [12, 14]. It is critical to establish an accurate crystal orientation with respect to the applied magnetic field in order to properly classify several centres having similar EPR parameters and all four fine structures belonging to each centre. Two sets of fine-structure lines became degenerate forming one set when the magnetic field **B** was applied in each of the crystallographic planes *ab*, *bc*, and *ca*. All four sets due to each of Cr^{3+} and Fe^{3+} at the four magnetically inequivalent sites merged completely into one when **B** was aligned along each of the crystal axes *a*, *b*, and *c*. This made it possible to establish crystal alignments within $\pm 0.05^{\circ}$ in the three planes by adjusting the crystal orientation inside the cylindrical cavity in such a way as to achieve degeneracy of the EPR lines [12, 14]. The EPR spectra were recorded by varying the orientations of **B** in the three planes from 0° to 180° in steps of 3° for the Fe³⁺ spectra.

3. Results and discussion

Depending on the value of the effective spin *S* of a paramagnetic ion and the symmetry around the substitutional ion, different numbers of the spin-Hamiltonian terms are required for a satisfactory description of the EPR spectra and their angular dependences [22]. For an Fe³⁺ ion with the effective spin S = 5/2 at a triclinic site in KTP, the appropriate spin Hamiltonian is given by [14, 15, 18]

$$H_{S} = \sum_{ij} \mu_{B} B_{i} g_{ij} S_{j} + \sum_{m=-2}^{2} B_{2}^{m} O_{2}^{m} + \sum_{m=-4}^{4} B_{4}^{m} O_{4}^{m}$$
(1)

where the first term represents the electron Zeeman interaction and the others the ZFS or the fine structure. Six independent elements for the **g**-tensor, five for the second-order ZFS tensor, and nine for the fourth-order ZFS tensor are needed to describe the Fe³⁺ spectra. On the order hand, for Cr^{3+} with S = 3/2 only the first two terms of equation (1) are needed [18, 19]. For all Cr^{3+} and Fe³⁺ spectra, the necessary spin-Hamiltonian parameters were determined [12–15] by employing a computer program, EPR-NMR [23]. The detailed procedure for calculating these parameters was described in [12, 14]. For each of the Cr^{3+} and Fe³⁺ centres, sixteen sets of fine structures were identified, and the sets were divided into four groups having the same principal values but different principal-axis orientations of **g** and the second-order ZFS tensors [12–15]. The groups were denoted as the centres A, B, C, and D for Cr^{3+} [12, 13], and C1, C2, C3, and C4 for Fe³⁺ [14, 15], each of which contains four fine structures arising from each of Cr^{3+} and Fe³⁺ at magnetically inequivalent sites.

For each of the fine structures, the program yields two sets of principal-axis orientations of each of the **g**-tensors and the second-order ZFS tensors represented in right-handed or left-handed coordinates, where each combination of them can be chosen as a set of principal axes. For comparison between the EPR data for Cr^{3+} and Fe^{3+} , all sets are now chosen to be represented in right-handed coordinates: the orientations of the **g**-tensors for SA1 and SB1 [12] and the second-order ZFS tensors for SA1 and SC1 [13] represented in left-handed coordinates are now converted into ones in right-handed-coordinate representations. The theta

Table 1. Principal-axis orientations of *g*-tensors and the second-order ZFS tensors with respect to the crystallographic axes (a, b, c) for Cr^{3+} in KTiOPO₄, and pairs of the principal axes having similar orientations of the two tensors for each centre and the angles between them.

	$A_{I}^{Cr}(SB1) \\$		$B_{I}^{Cr}(SD1)$		$C_{I}^{Cr}(SA1)$		$D_{I}^{Cr}(SC1)$					
	Orientations (deg) of the principal axes of <i>g</i> -tensors ^a											
	θ	ϕ	θ	ϕ	θ	ϕ	θ	φ				
Ox'	131	349	142	350	103	86	63	217				
Oy'	139	183	119	214	14	107	139	271				
<u>Oz'</u>	97	85	112	111	85	357	62	323				
	Orientations (deg) of the principal axes of the second-order ZFS tensors ^b											
	θ	ϕ	θ	ϕ	θ	ϕ	θ	φ				
Ox	53.88	350.70	51.47	354.95	67.48	261.63	50.96	268.35				
Oy	143.72	345.64	41.17	346.53	157.45	265.07	140.94	270.66				
Oz	92.89	79.57	94.12	81.66	88.78	352.14	88.87	359.27				
	Pairs and angles (deg) ^a											
	(Oz', Oz); 7		(O <i>x</i> ′, O <i>y</i>); 3		(Oz', Oz); 6		(Oy', Oy); 2					

^a The estimated maximum uncertainties in the angle are $\pm 2.6^{\circ}$ for all centres.

^b The estimated maximum uncertainties in the angle are $\pm 0.05^{\circ}$ for all centres.

values of the principal axes of the **g**-tensor are x' for SC1 and z' for SD1 (with the typographical errors in [13] corrected). After comparison of the principal-axis orientations among 16 fine structures, a representative fine structure having the closest principal-axis orientations to the other three representative ones have been chosen for each centre. After overall conversion, the principal-axis orientations of the **g**-tensors and the second-order ZFS tensors with respect to the crystallographic axes for the Cr³⁺ centres are listed in table 1, where the centres B, D, A, and C from our previous reports [12, 13] are renamed as A^{Cr}, B^{Cr}, C^{Cr}, and D^{Cr}, respectively. The subscripts, i = I, II, III, and IV, indicate the fine structures due to the Cr³⁺ substituted at four magnetically inequivalent Ti sites, I, II, III, and IV, respectively [12–15]. The line intensities of A^{Cr} and C^{Cr} are much stronger than those of B^{Cr} and D^{Cr} [13].

The orientations of the principal axes x, y, and z of the second-order ZFS tensors for A^{Cr} and B^{Cr} nearly coincide with those of the Ti(1)–O(2), Ti(1)–O(1), and Ti(1)–O(6) bonds, respectively, and those for C^{Cr} and D^{Cr} nearly coincide with those of the bonds Ti(2)–OT(1) (rather than Ti(2)–OT(2) as given in [13]), Ti(2)–O(8), and Ti(2)–O(4), respectively. From this result it was concluded that the centres A^{Cr} and B^{Cr} as well as C^{Cr} and D^{Cr} with similar principal-axis orientations originate from Cr^{3+} at the sites Ti(1) and Ti(2), respectively [13]. For each centre, one principal-axis direction of the **g**-tensors is nearly the same as one of those of the second-order ZFS tensor. For the two quantities, the pairs of similar principal axes of **g** and the second-order ZFS tensors are as follows: z' and z for A_1^{Cr} and C_1^{Cr} , x' and y for B_1^{Cr} , and y' and y for D_1^{Cr} as listed in table 1.

The principal-axis orientations of Fe³⁺ centres have been chosen like those of Cr³⁺: the orientations of the second-order ZFS tensors of C1 and C3, named in [14, 15], have now been chosen from another set of two yielded by the computer program [23]. The principal-axis orientations of the second-order ZFS tensors with respect to the crystallographic axes for the Cr³⁺ and the Fe³⁺ centres are listed in table 2, where the centres called C1, C3, C2, and C4 in [14, 15] are renamed as A^{Fe}, B^{Fe}, C^{Fe}, and D^{Fe}, respectively. The line intensities of A^{Fe} and C^{Fe} are much stronger than those of B^{Fe} and D^{Fe}, like those of the Cr³⁺ centres. The

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Table 2. Principal values of the second-order ZFS tensors and their principal-axis orientations with respect to the crystallographic axes (a, b, c) for Cr^{3+} and Fe^{3+} in KTiOPO₄, and pairs of the principal axes having similar orientations of Cr^{3+} and Fe^{3+} centres and the angles between them.

	Cr ³⁺ centres										
	A	Cr(SB1)	В	^{Cr} (SD1)	C	Cr ₁ (SA1)	E	O ^{Cr} (SC1)			
	Principal values of the second-order ZFS tensors ^a										
$\overline{B_2^0}_{(cm^{-4})}$	1365.4 1194.9		1259.4 1227.7			1675.7 932.4		1451.4 1361.1			
B_2^2 (cm ⁻⁴)											
N_2 (cm ⁻⁸)	2	2340246		2088504		3097760		2724093			
		Orientatio	ons (deg) o	of the principal	axes of the	second-order Z	FS tensors	b			
	θ	φ	θ	φ	θ	ϕ	θ	ϕ			
Ox	53.88	350.70 345.64	51.47	354.95 346 53	67.48 157.45	261.63	50.96	268.35 270.66			
Oz	92.89	79.57	94.12	81.66	88.78	352.14	88.87	359.27			
				Fe ³⁺	centres						
	A _I ^{Fe} (C1b)		B _I ^{Fe} (C3b)		(C _I ^{Fe} (C2a)		D _I ^{Fe} (C4a)			
			Principa	l values of the	second-ord	ler ZFS tensors ^a					
$\frac{B_2^0}{(cm^{-4})}$	-817.3		-819.3		_	-739.1		-722.5			
B_2^2 (cm ⁻⁴)	-443.1		-439.8		-442.2		-552.9				
N_2 (cm ⁻⁸)	733425		735727		611449		623906				
		Orientatio	ons (deg) o	of the principal	axes of the	second-order Z	FS tensors	b			
	θ	ϕ	θ	ϕ	θ	ϕ	θ	ϕ			
Ox Oy Oz	66.23 132.45 51.83	260.98 327.22 11.23	65.88 132.68 51.91	260.70 326.54 17.17	67.84 140.02 58.65	258.15 319.09 2.51	63.26 138.00 60.31	259.43 315.41 6.13			
	Pairs and angles (deg) ^b										
	$\begin{array}{c} \mathbf{A}_{\mathrm{I}}^{\mathrm{Cr}}\left(\mathrm{O}x\right),\mathbf{C}_{\mathrm{I}}^{\mathrm{Fe}}\left(\mathrm{O}z\right);\\ 10.9\end{array}$		$B_{\rm I}^{\rm Cr} ({\rm O}x), C_{\rm I}^{\rm Fe} ({\rm O}z);$ 9.4		C _I ^{Cr} (C	$C_{I}^{Cr} (Ox), A_{I}^{Fe} (Ox);$ 1.4		$D_{\rm I}^{\rm Cr}$ (Ox), $A_{\rm I}^{\rm Fe}$ (Ox); 16.5			
	$\begin{array}{c} \mathrm{A}_{\mathrm{I}}^{\mathrm{Cr}}\left(\mathrm{O}x\right),\mathrm{D}_{\mathrm{I}}^{\mathrm{Fe}}\left(\mathrm{O}z\right);\\ 14.4\end{array}$		$B_{I}^{Cr} (Ox), D_{I}^{Fe} (Oz);$ 12.3		$C_{\rm I}^{\rm Cr} ({\rm O}x), {\rm B}_{\rm I}^{\rm Fe} ({\rm O}x);$ 1.8		D_{I}^{Cr} (Ox), B_{I}^{Fe} (Ox); 16.5				
	—				C _I ^{Cr} (C	$\begin{array}{c} C_{\mathrm{I}}^{\mathrm{Cr}}\left(\mathrm{O}x\right), C_{\mathrm{I}}^{\mathrm{Fe}}\left(\mathrm{O}x\right);\\ 3.3\end{array}$		_			
	—		_		C _I ^{Cr} (C	$C_{I}^{Cr} (Ox), D_{I}^{Fe} (Ox);$ 4.7		_			

 a The estimated maximum uncertainties in the angle are $\pm 0.3^\circ$ for all centres.

 a The estimated maximum uncertainties in the angle are $\pm 0.05^\circ$ for all centres and all pairs.

representative fine structures A_I^{Fe} and B_I^{Fe} correspond to C1b and C3b. For a paramagnetic ion with $S \ge 5/2$ at a triclinic site, the symmetry of the fourth-order ZFS terms mainly reflects that of the nearest neighbours around the substitutional-ion sites [24, 25]. Therefore, in this case, unlike that of Cr^{3+} , a comparison of the orientations of the principal axes between the secondorder ZFS tensors and the Ti–O bonds cannot provide an indicator for identifying the origin of the centres. By comparing the result [18] obtained from pseudosymmetry-axis method [24, 25] and our previous results [14, 15], the centres A^{Fe} and B^{Fe} were found to arise from Fe³⁺ at Ti(1), and C^{Fe} and D^{Fe} from Fe³⁺ at Ti(2). Unlike the case for Cr^{3+} , the **g**-tensor and the second-order ZFS tensor of the Fe³⁺ centre do not have any similar principal-axis directions. For each centre, however, one principal-axis direction of the second-order ZFS tensor of Cr^{3+} is nearly the same as one of those of Fe³⁺. For the two quantities, the pairs of similar axes are listed in table 2.

Since the norm defined in [22] is invariant under arbitrary rotations of the frame of coordinates, it is useful to examine the equivalency of the centres represented in various coordinate systems. In addition to the norm, an investigation of the closeness of the mutual orientations of two vectors measured in terms of the normalized scalar product of the principalaxis orientations of the second-order ZFS tensors is also useful, for examining the equivalency [26]. From the closeness of the numerically defined norm and the mutual orientation, it was concluded in [26] that the two centres A^M and B^M are equivalent. However, in the case of the four Cr³⁺ and the four Fe³⁺ centres in KTP with very similar orientations of the secondorder ZFS tensors, the two numerical definitions are not sufficient to allow us to reach such a conclusion. One of the most reliable pieces of evidence for the non-equivalency of the four centres can be obtained by a consideration of the following two facts: (i) A^{M} and C^{M} have conspicuously stronger line intensities than B^M and D^M [12-15], and (ii) the four complete sets of fine-structure lines arising from Cr³⁺ and Fe³⁺ are observed at magnetically inequivalent sites in the skew planes which deviate slightly from the three crystallographic planes. On the basis of this evidence we have reached the conclusion that the four Cr^{3+} and the four Fe^{3+} centres are definitely distinct.

As an explanation as to why such kinds of the two different centres arise from the crystallographically equivalent Ti sites, a vacancy of the first-neighbour oxygen OT(1) was suggested as a divalent positive charge compensator simultaneously for the two Cr^{3+} ions and the two Fe^{3+} ions at the adjacent Ti(1) and Ti(2) sites [13, 15]. In this case, the closeness of the second-order ZFS tensor data sets for the two Cr^{3+} centres and two Fe^{3+} centres would be hard to understand, since the vacancy would force the original Ti–O orientations to be substantially deformed. As a result, the two centres would have quite different orientations of the principal axes of the second-order ZFS tensors and their principal values. Consequently, this type of charge compensator is more likely to be placed beyond the first neighbour around the substitutional Cr^{3+} and Fe^{3+} ions.

The electron-trapping Cr^{5+} centres and hole centres $[HCrO_3]^-$ were observed at about 173 K in chromium-doped KTP after it had been irradiated with x-rays at room temperature [27], and Cr^{6+} was found to be substituted for P^{5+} in a slightly distorted tetrahedron, $P^{5+}(O^{2-})_4$. In addition to this result, a more probable origin of the four centres than previous ones [13, 15] is illustrated in figure 1. The ionic radius of Ti^{4+} (0.68 Å) is very similar to that of Cr^{3+} (0.63 Å), but the charge states of the two ions are different. When Cr^{3+} ions substitute for Ti sites, these defect centres should be compensated by a monovalent charge for each of the Cr^{3+} ions to achieve electrical neutrality. A Cr^{6+} ion substituted for P^{5+} may be a good monovalent charge compensator. The original P–O bond orientations in the PO₄ octahedron may be somewhat deformed by the substitutional Cr^{6+} ion, with its ionic radius (0.52 Å) being larger than that of P^{5+} (0.35 Å); however, the original Ti–O structures may not be disturbed



Figure 1. Projection of the KTP structure on the *ca*-plane, and the charge-compensation models. The substitutional M^{3+} (M = Cr or Fe) sites and Cr^{6+} sites playing the role of charge compensators are as follows: (a) M^{3+} at Ti(1), Ti⁴⁺ at Ti(2), and Cr^{6+} at P(1) or P(2) for A^M; (b) Ti⁴⁺ at Ti(1), M^{3+} at Ti(2), and Cr^{6+} at P(1) or P(2) for C^M; and (c) M^{3+} simultaneously at Ti(1) and Ti(2) as well as Cr^{6+} simultaneously at P(1) and P(2) for both centres B^M and D^M.

very much by this charge compensator. Such Cr^{6+} ions can play the role of monovalent charge compensators with little deformation of the Ti–O bond orientations as shown in figure 1. The probability of Cr^{3+} ions being substituted simultaneously at both Ti sites (case (c)) might be much less than that of them being substituted at either Ti(1) or Ti(2) sites. The centres A^M and C^M with comparatively much stronger intensities appear to be originating from the more abundant Cr^{3+} ions in cases (a) and (b). Nevertheless, there are two possibilities for assigning a certain site of the charge compensator Cr^{6+} to A^M and C^M using the present data alone. This charge-compensation model shown in figure 1 can be applied to Fe³⁺ (0.64 Å) centres on the basis of the fact that Cr^{3+} spectra are also present for the Fe-doped KTP crystal.

In summary, it has been found that the two numerically defined norms [22] and the normalized scalar product of the principal-axis orientations of the second-order ZFS tensors [26] are not very useful for investigating the equivalency of the four distinct Cr^{3+} centres and the four Fe³⁺ centres in KTP. It has been found that one principal-axis direction of the **g**-tensor is nearly the same as one of those of the second-order ZFS tensor for Cr^{3+} centres, and one

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principal-axis direction of the second-order ZFS tensor for Cr^{3+} centres is also similar to one of those for the Fe³⁺ centres. A Cr^{6+} ion substituted for P⁵⁺ in a PO₄ tetrahedron plays the role of a positive monovalent charge compensator to achieve electrical neutrality when Cr^{3+} and Fe³⁺ ions substitute at Ti(1) or Ti(2) sites. The similar orientations of the principal axes of the second-order ZFS tensors of A^M and B^M as well as C^M and D^M, arising from M (M = Cr or Fe) at the respective Ti(1) and Ti(2) sites, are well explained by the suggested charge-compensation models in connection with the Cr⁶⁺ ions substituting for P⁵⁺ ions.

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